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An Efficient Oxidizing Reagent for the Synthesis of Mixed Backbone Oligonucleotides via the H-Phosphonate Approach

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Abstract—The mixture of carbon tetrachloride, *N*-methyl morpholine (NMM), pyridine and water in acetonitrile has been exploited for the oxidation of dinucleoside H-phosphonate diesters to the corresponding phosphates. The system is found to be inert to the phosphoramidate (P-N) and the phosphorothioate (P-S) linkages and has successfully been applied to the solid phase synthesis of mixed-backbone oligonucleotides (MBOs).

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Introduction

The synthesis of oligonucleotides by the phosphoramidite¹ and the H-phosphonate² methodologies has accelerated the development of oligonucleotides as potential therapeutics and diagnostic agents and for the applications in the field of molecular biology. A significant potential for success to control the natural cellular functions, is the use of the synthetic oligonucleotides with modified backbones that are complementary to the gene or mRNA responsible for those functions. Some of the important examples of the phosphate modifications being investigated are the phosphorothioates, phosphoramidates and phosphotiesters. These modifications in the oligonucleotide can lead to enhanced cell membrane permeability, increased resistance to exo and/or endonuclease activity and increased stability of oligonucleotide/targeted sequence duplex. The nucleoside H-phosphonates provide a convenient route to these types of modifications with the advantage that a common intermediate would be required to obtain any of the various modifications.

The synthetic cycle for the H-phosphonate chemistry is relatively simple as compared to the phosphoramidite approach since the oxidation step is carried out at the end of the synthesis, before the final detritylation. The efficiency of the oxidation is therefore a highly critical

issue. The oxidation procedure with the 0.1 M I₂ in Py/H₂O/THF (5/5/90) provides inconsistent oxidation of the internucleotide H-phosphonate linkages.³ Results have been improved by using a two step oxidation, beginning with iodine solution containing a weak base like *N*-methyl imidazole followed by a solution containing a strong base like triethyl amine.³ Additionally, iodine solution containing water loses the ability to oxidize support bound oligo H-phosphonate after a couple of days. Hence mixing the fresh iodine solution prior to oxidation is advised.³ An alternative method for oxidation is a solution of 2% iodine in Py/H₂O (98:2,v/v).⁴ However this oxidation strategy by aqueous iodine in pyridine is problematic because the aqueous condition brings about partial hydrolysis of the internucleotidic H-phosphonate linkage.³ de Vroom et al.⁵ has developed an alternative method of oxidation using *tert*-butyl hydroperoxide (TBHP) in the presence of *N,O*-bis(trimethylsilyl) acetamide (BSA). In this case BSA, not only activates the nucleoside H-phosphonate but also acts as a scavenger of all traces of water in the oxidation reaction. However the excess use of BSA, converts TBHP to the less reactive trimethylsilyl *tert*-butyl hydroperoxide, and consequently causes retarding of the oxidation reaction.

As the oligonucleotide-based applications continue to evolve, synthetic methodologies need to be refined to meet the needs. The use of phosphorothioate oligodeoxynucleotides (PS-oligos), has resulted in the first antisense drug Fomivirsen⁶ (vitravene). However, these PS-oligos are not optimal antisense molecules and their

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drawbacks have been revealed in the development process of the drugs.⁷ Therefore at the current stage of technology, the mixed backbone oligonucleotides (MBOs) are considered as the best choice as the second generation antisense compounds.⁸

As part of our interest in the field of oligonucleotide synthesis⁹ we have optimized $\text{CCl}_4/\text{H}_2\text{O}/\text{NMM}/\text{Py}/\text{CH}_3\text{CN}$ ¹⁰ oxidizing mixture to the H-phosphonate chemistry and in turn to the synthesis of MBOs containing the phosphorothioate and phosphoramidate linkages. The system is efficient and fast in converting nucleoside H-phosphonate diesters to the corresponding

phosphates both in the solution phase and the solid phase synthesis of oligonucleotides.

Results and Discussion

In order to explore the role of $\text{CCl}_4/\text{H}_2\text{O}/\text{NMM}/\text{Py}/\text{CH}_3\text{CN}$ as an oxidizing reagent in the H-phosphonate chemistry, preliminary experiments were carried out in the solution phase. The unoxidized dithymidyl H-phosphonate diester **1** (0.01 mmol) ($\delta = 12.1$ ppm) was subjected to oxidation with $\text{CCl}_4/\text{H}_2\text{O}/\text{NMM}/\text{Py}/\text{CH}_3\text{CN}$ (2.5/1.0/1.0/6.0/1.0, v/v) (Scheme 1). The addition of $\text{CCl}_4/\text{H}_2\text{O}/$

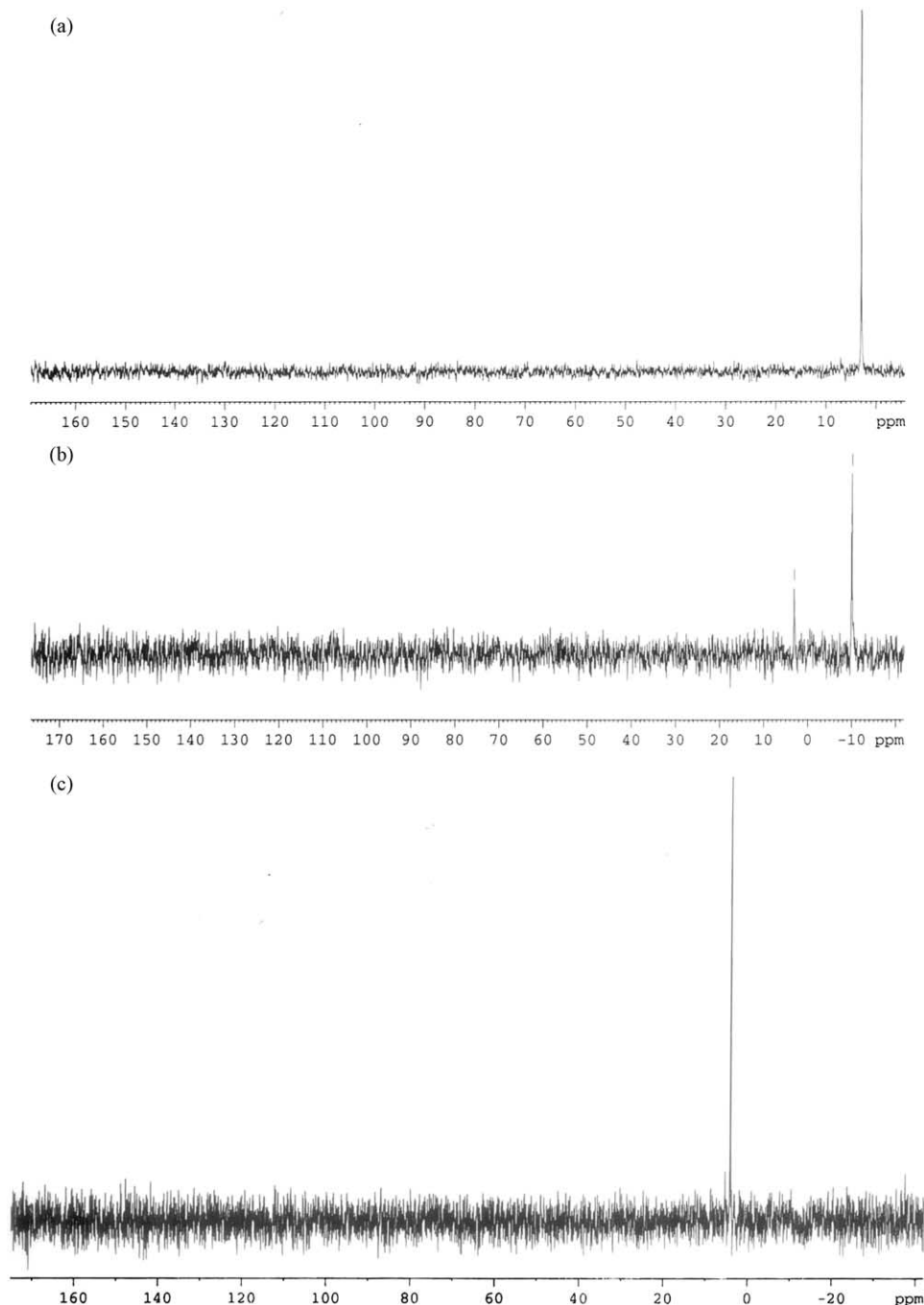


Figure 1. ^{31}P NMR spectra of the dithymidyl phosphate, **2** synthesized by using: (a) $\text{CCl}_4/\text{H}_2\text{O}/\text{NMM}/\text{Py}/\text{CH}_3\text{CN}$ (2.5/1.0/1.0/6.0/1.0, v/v), (b) with 0.035 M I_2 in $\text{Py}/\text{THF}/\text{H}_2\text{O}$ (1/300/33, v/v), (c) 0.2 M iodine in $\text{Py}/\text{H}_2\text{O}$ (98:2, v/v).

NMM/Py/CH₃CN to the unoxidized dithymidyl H-phosphonate diester resulted in almost immediate (<5 min) and quantitative conversion to dithymidyl phosphate **2** (δ =2.9 ppm), with no detectable side products as judged by ³¹P NMR spectroscopy (Fig. 1(a)). A similar experiment was performed with 0.035 M I₂ in Py/THF/H₂O (1/300/33, v/v). The oxidation reaction in this case did not lead to complete formation of **2**. Two products were formed, one of them was the phosphate

(3.0 ppm), and the other a side product (δ =−9.8 ppm) (Fig. 1(b)).

However, when 0.2 M I₂ in Py/H₂O (98:2,v/v) was added, the oxidation was immediate (<5 min) and quantitative, with no detectable side products (Fig. 1C). These results indicate that the same combination of 0.035 M I₂ in Py/THF/H₂O (1/300/33, v/v) which is conventionally used in the phosphoramidite approach,

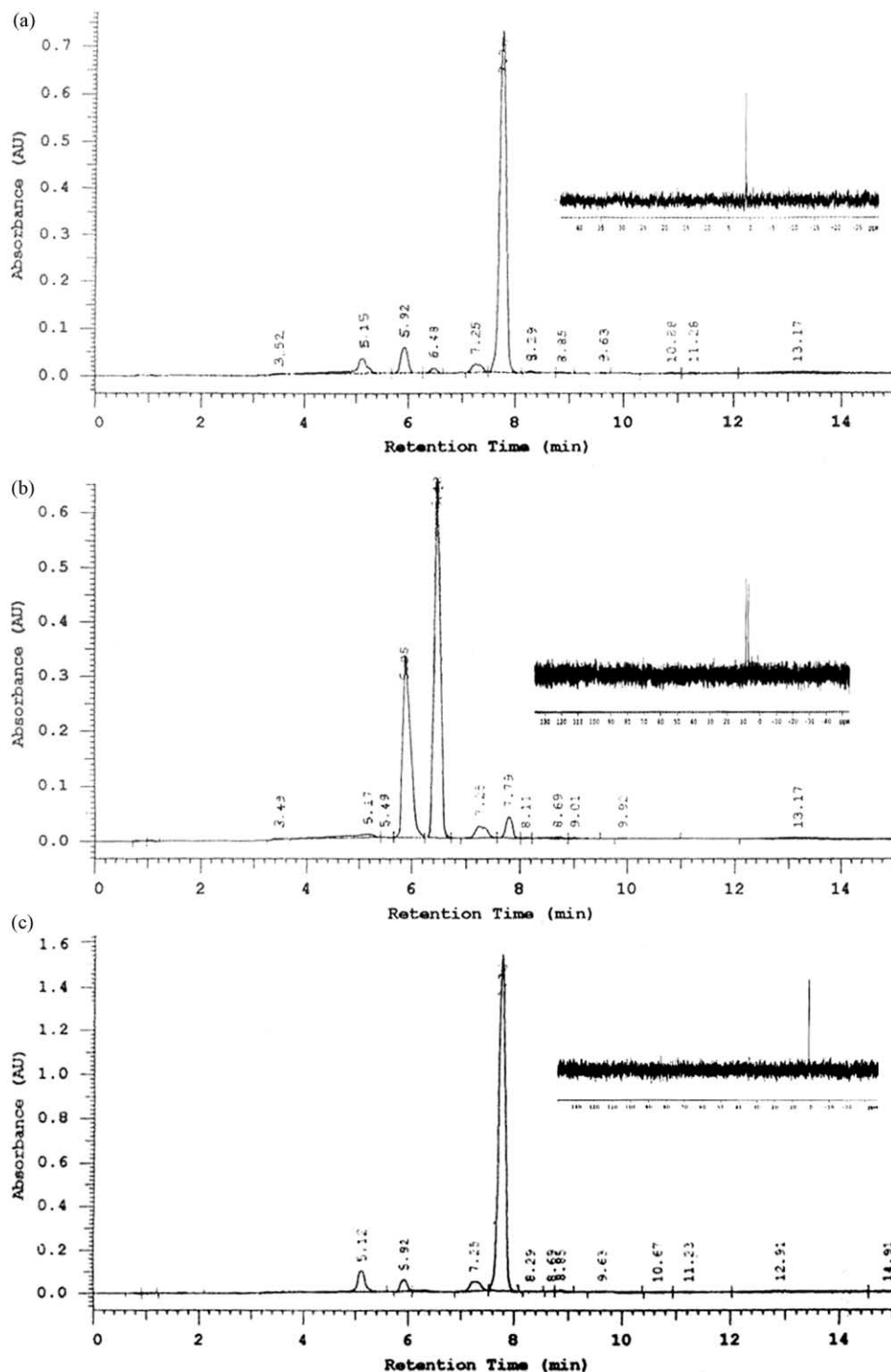


Figure 2. RP-HPLC (³¹P NMR is shown in the inset) of d(AA), **3** synthesized by: (a) I₂ in Py/H₂O method (positive control), (b) no oxidation method (negative control), (c) CCl₄/H₂O/NMM/Py/CH₃CN method.

cannot be used as an oxidant in the H-phosphonate chemistry which demands 0.2 M iodine in Py/H₂O (98:2, v/v). In contrast, the CCl₄/H₂O/NMM/Py/CH₃CN (2.5/1.0/1.0/6.0/1.0, v/v) mixture can be used as a versatile oxidizing reagent for the solution phase oxidation of nucleoside phosphite triester to the corresponding phosphotriester via the phosphoramidite approach¹⁰ as well as for the conversion of the nucleoside H-phosphonate diesters to corresponding phosphodiester via the H-phosphonate approach.

In order to check the suitability of the developed CCl₄/H₂O/NMM/Py/CH₃CN mixture on solid phase, the dimer d(AA) (**3**) was synthesized by coupling of dA^{bz} H-phosphonate with dA^{bz} lcaa CPG 500 Å using the

standard protocols for the H-phosphonate chemistry. The oxidation was effected by the CCl₄/H₂O/NMM/Py/CH₃CN (2.5/0.2/1.0/6.0/9.3, v/v) system for 45 min.¹¹ In order to check the oxidizing efficiency of the system, control experiments were performed. One control experiment (positive control) involved the oxidation of the unoxidized d(A_{pH}A) dimer supported on lcaa CPG-500 Å with the conventionally used 0.2 M I₂ in Py/H₂O (98:2, v/v) for 30 min. In the second control experiment (negative control), the d(A_{pH}A) dimer supported on lcaa CPG-500 Å was not oxidized at all. After performing the standard deprotection under basic conditions, the crude deprotection mixtures of the three experiments were analyzed by RP-HPLC and ³¹P NMR spectroscopy and were compared with each other. The

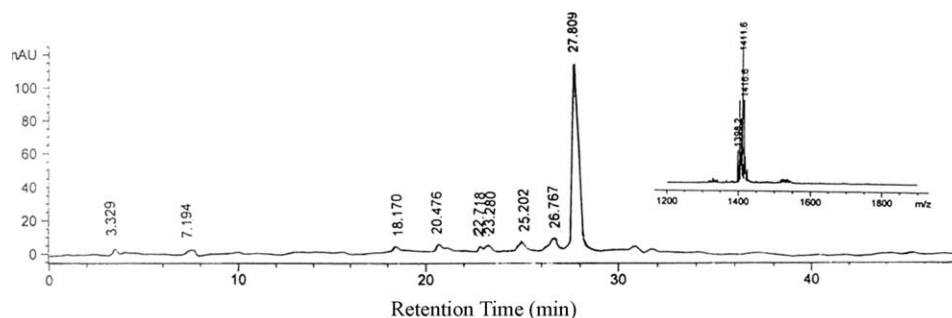


Figure 3. RP-HPLC profile (ESI-MS shown in the inset) of d(GTTAAGACTTTTAC), **4** synthesized by CCl₄/H₂O/NMM/Py/CH₃CN method. ESI-MS of **4** *MW*_{found}: 4234.8 *MW*_{calcd}: 4237.

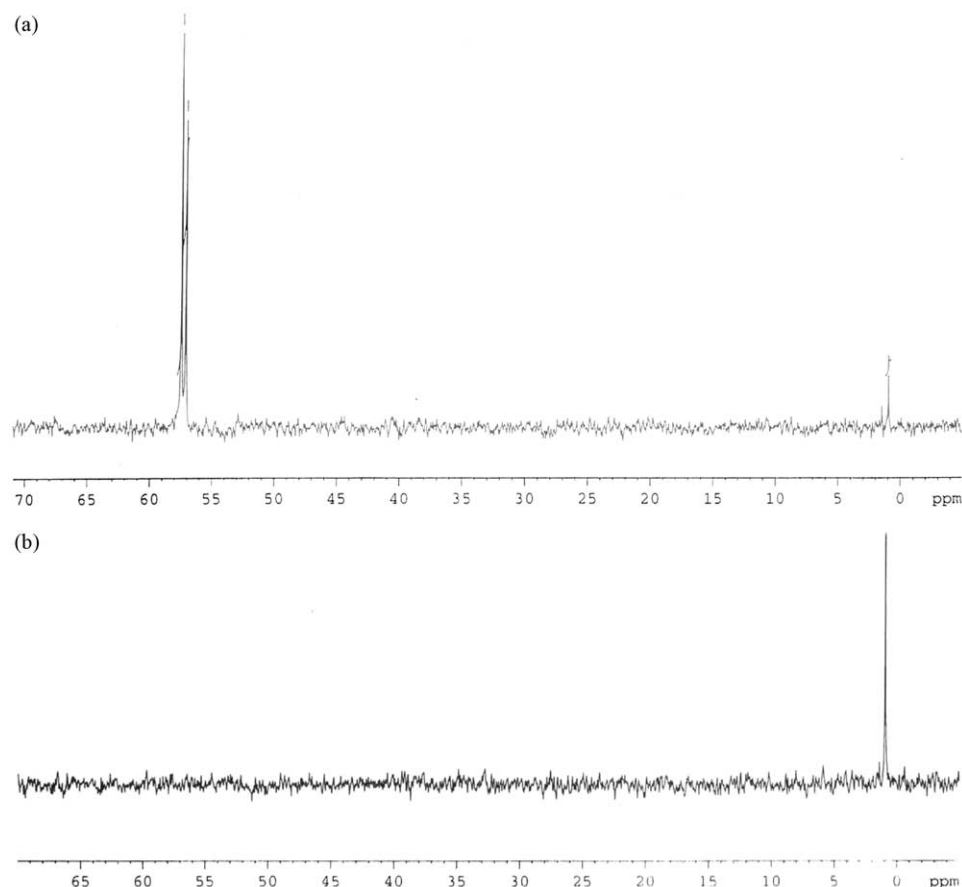


Figure 4. Solution phase stability of d(TpsT), **5** in: (a) CCl₄/H₂O/NMM/Py/CH₃CN system and (b) I₂ in Py/H₂O system.

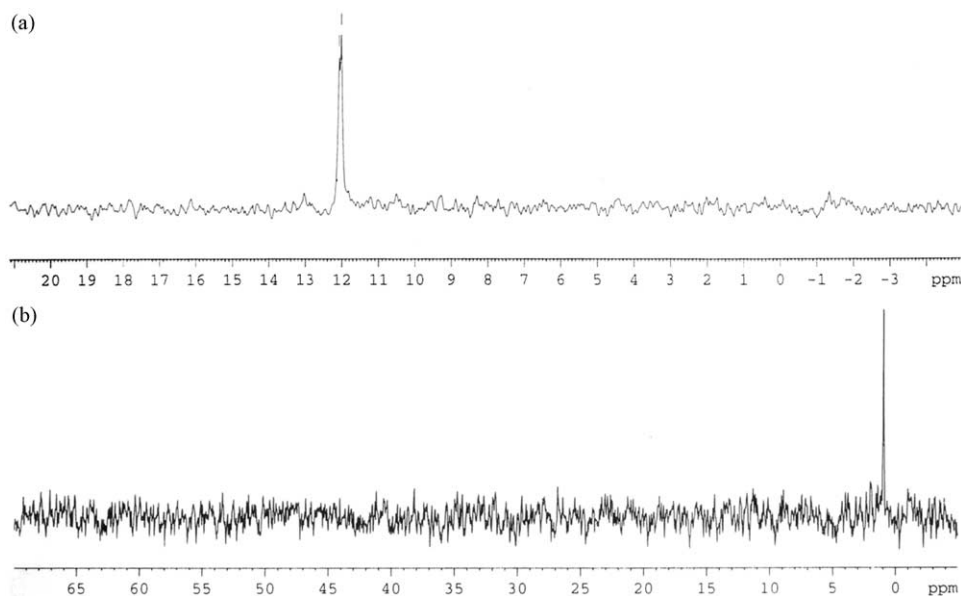


Figure 5. Solution phase stability of d(T_{pN}T) **6**, in: (a) CCl₄/H₂O/NMM/Py/CH₃CN system and (b) I₂ in Py/H₂O system.

RP-HPLC analysis of the **3** synthesized by I₂ in Py/H₂O showed the oxidized dimer **3**, as the major peak at retention time 7.76 min, its ³¹P NMR showed signal at $\delta = 0.93$ ppm arising due to the phosphodiester linkages (Fig. 2A). The negative control showed two peaks one at retention time 5.95 min (adenosine) and the other at retention time 6.48 min arising due to the formation of H-phosphonate, with <3% formation of **3** (retention time 7.79 min), while its ³¹P NMR showed two signals at $\delta = 6.9$ and 8.3 ppm (Fig. 2B). The RP-HPLC analysis of the crude deprotection mixture by CCl₄/H₂O/NMM/Py/CH₃CN method, showed the major peak at retention time 7.76 min (expected product **3**), with no evidence for the formation of the H-phosphonate arising due to the cleavage of the unreacted H-phosphonate diester linkages under basic conditions used to release the oligonucleotide from the solid support (Fig. 2(c)). Also, ³¹P NMR of its crude deprotection mixture indicated only one signal at $\delta = 0.9$ ppm (shown in the inset of Fig. 2C). These experiments clearly indicate that the conversion of oligonucleotide H-phosphonate diesters to the corresponding phosphates results in quantitative conversion in case of the optimized CCl₄/H₂O/NMM/Py/CH₃CN system. In case of I₂ in Py/H₂O system ca. 2% formation of H-phosphonate (retention time 6.4 min) was detected. It has been reported that the oxidation of H-phosphonate with the I₂ in Py/H₂O system is slow and it is preferable to do it in an automated mode using a three reagent combination in a DNA synthesizer.¹²

The oxidizing system has successfully been applied to the synthesis of 14 mer oligo d(GTTAAGACTTTTAC) (**4**). The RP-HPLC profile of the crude deprotection mixture of **4** is shown in Figure 3. All the fractions were characterized simultaneously by ESI-MS analysis (ESI-MS of the major fraction, **4** is shown in the inset of Fig. 3).

In order to assess the chemical stability of the CCl₄/H₂O/NMM/Py/CH₃CN system towards the phosphorothioate

linkages in the backbone, the dimer d(TpsT) (**5**) (1.0 A₂₆₀ units) ($\delta = 57.0$ and 57.4 ppm) was stirred for 16 h with CCl₄/H₂O/NMM/Py/CH₃CN (2.5/1.0/1.0/6.0/1.0, v/v) mixture at ambient temperature. A similar experiment was carried out under identical scale, conditions and protocols using 0.2 M I₂ in Py/H₂O (98:2, v/v). In order to judge the potential occurrence of desulfurization by these oxidizing reagents, the ³¹P NMR analysis of the reaction mixtures were carried out. The ³¹P NMR of the crude reaction mixture using CCl₄/H₂O/NMM/Py/CH₃CN reagent (Fig. 4(a)) indicated ca. 6% desulfurization $\delta = 0.9$ ppm). However, the crude reaction mixture using I₂ in Py/H₂O (98:2, v/v) method indicated complete desulfurization of **5**, to give d(TpoT) as the only detectable product ($\delta = 0.96$ ppm) (Fig. 4(b)) as judged by the ³¹P NMR analysis. The desulfurization process by I₂ in Py/H₂O has been reported previously in literature.¹³

Similarly in order to check the compatibility of the CCl₄/H₂O/NMM/Py/C₃CN system towards the phosphoramidate linkages in the backbone, the dimer d(T_{pN}T)(**6**) (1.0 A₂₆₀ units) = 12.08 and 12.00 ppm) was stirred individually for 16 h with the CCl₄/H₂O/NMM/Py/CH₃CN (2.5/1.0/1.0/6.0/1.0, v/v) and 0.2 M I₂ in Py/H₂O (98:2, v/v) oxidizing reagents, as described previously. The ³¹P NMR analysis of the crude reaction mixtures were then carried out. The dimer **6** stirred with the CCl₄/H₂O/NMM/Py/CH₃CN did not show any cleavage of the phosphoramidate linkages as judged by the phosphorus NMR (Fig. 5(a)). However, the phosphoramidate linkages were completely cleaved by I₂ in Py/H₂O reagent to give the phosphodiester signal at $\delta = 0.97$ ppm which was the only detectable product (Fig. 5(b)). These solution phase analysis clearly indicate that the CCl₄/H₂O/NMM/Py/CH₃CN oxidation method is far superior than the conventionally used I₂ in Py/H₂O in retaining the phosphorothioate and the phosphoramidate linkages in the oligonucleotide backbone.

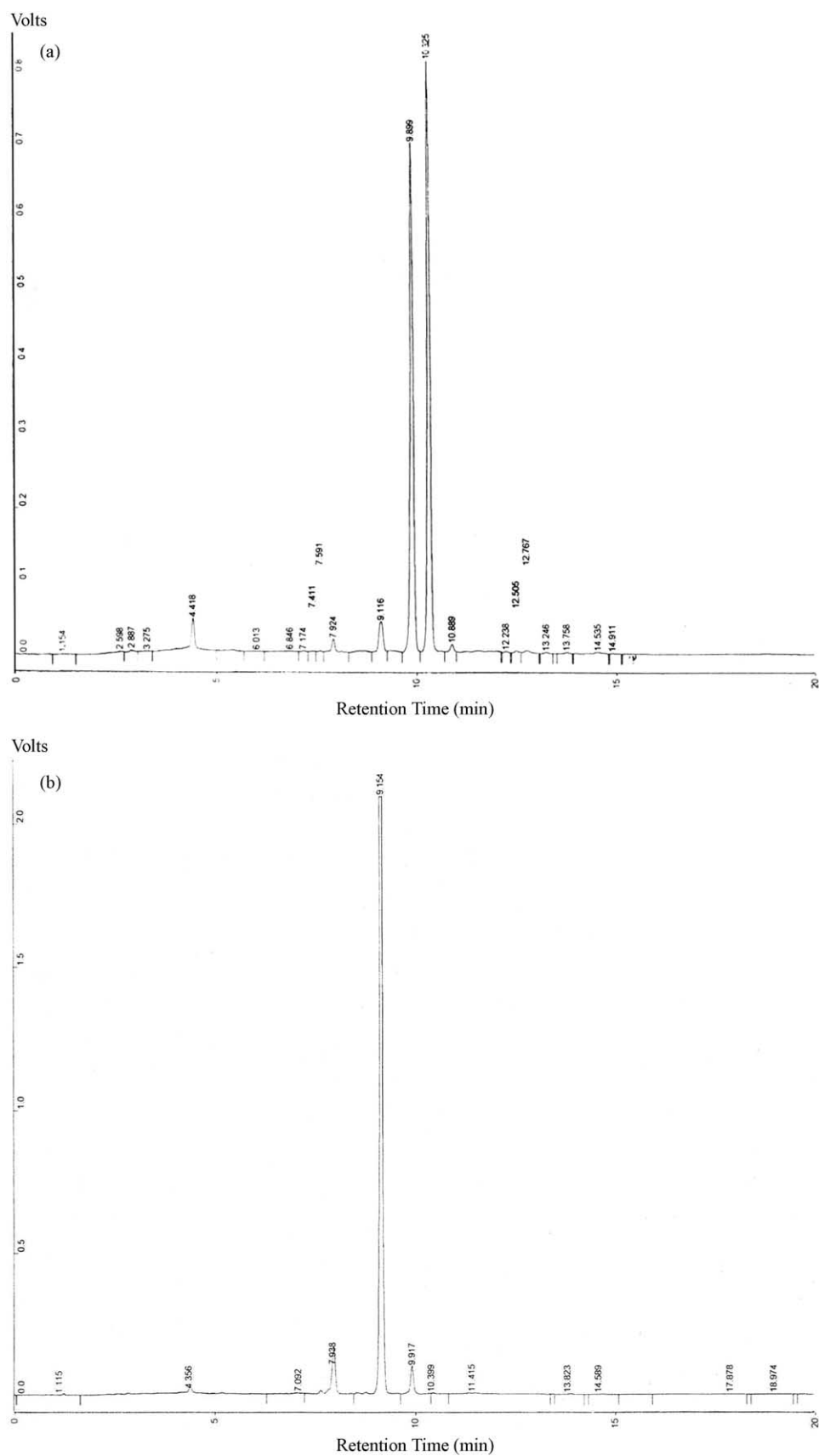


Figure 6. RP-HPLC profile of d(TpoTpsT), **7** synthesized by: (a) CCl₄/H₂O/NMM/Py/CH₃CN method and (b) I₂ in Py//H₂O method.

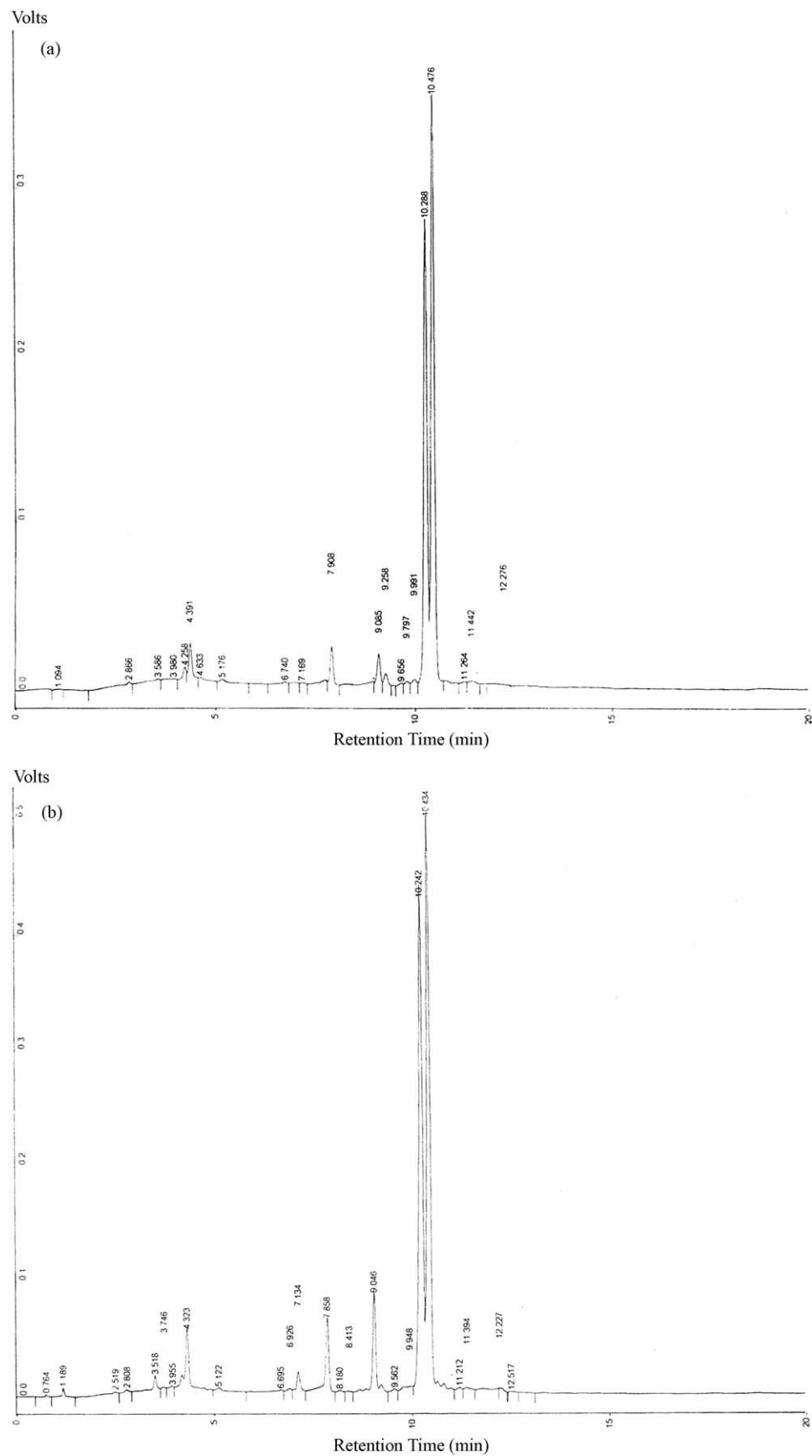


Figure 7. RP-HPLC profile of d(TpoT_{PN}T), **7** synthesized by: (a) CCl₄/H₂O/NMM/Py/CH₃CN method and (b) I₂ in Py/H₂O method.

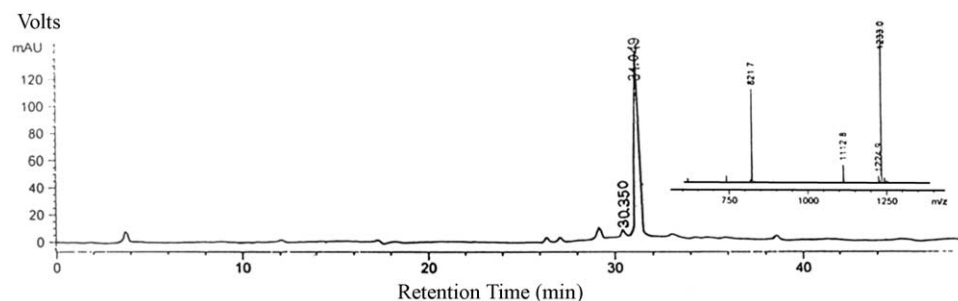
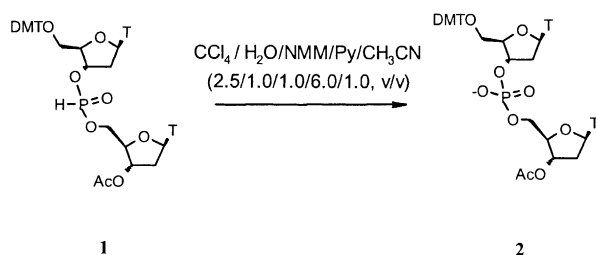
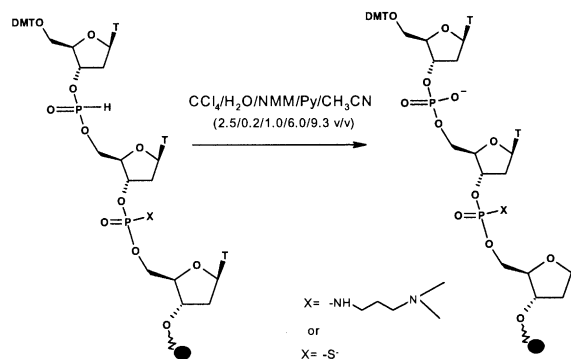


Figure 8. RP-HPLC profile (ESI-MS shown in the inset) of d(Tp_oTp_sTp_sTp_sTp_sTp_sTp_sT), **9** synthesized by CCl₄/H₂O/NMM/Py/CH₃CN method. ESI-MS of **9** MW_{calcd} : 2459.95, MW_{found} : 2466.



Scheme 1. Solution phase oxidation of dithymidyl H-phosphonate diester **1** with CCl₄/H₂O/NMM/Py/CH₃CN system.



Scheme 2.

Encouraged by these positive results on the dimers **5** and **6** we decided to investigate the use of the CCl₄/H₂O/NMM/Py/CH₃CN method to the synthesis of trimers d(Tp_oTp_sT) (**7**) and d(Tp_oP_NT) (**8**) [where PN corresponds to P-NH-(CH₂)₃N(CH₃)₂ linkages] on solid phase using the standard protocols for the H-phosphonate chemistry (**Scheme 2**). After the standard deprotection step, the crude deprotection mixtures were analysed by RP-HPLC. The RP-HPLC of **7** synthesized by CCl₄/H₂O/NMM/Py/CH₃CN (2.5/0.2/1.0/6.0/9.3, v/v) method showed ca. 2% desulfurization (retention time 9.11 min), while the expected product **7** was obtained as the major peak at retention time 9.8 and 10.35 min (two peaks are due to the diastereomers resulting from the chiral phosphorus centre) (**Fig. 6(a)**). The RP-HPLC of **7** synthesised by 0.2 M I₂ in Py/H₂O (98:2, v/v) method however resulted in nearly complete loss of sulfur to give the desulfurized trimer, d(Tp_oTp_oT) as the major peak at retention time 9.15 min. The expected product **7** obtained was ca. 4% (**Fig. 6(b)**).

Similarly the trimer **8** was synthesized by using CCl₄/H₂O/NMM/Py/CH₃CN (2.5/0.2/1.0/6.0/9.3, v/v) and 0.2 M I₂ in Py/H₂O (98:2, v/v) method on solid phase. The RP-HPLC analysis of the crude deprotection mixtures were then carried out. The CCl₄/H₂O/NMM/Py/CH₃CN method, although was stable towards the phosphoramidate linkages in the solution phase, in the solid phase cleaved ca. 2% of the phosphoramidate linkages (retention time 9.0 min), while **8** was obtained as major peak (retention time 10.2 and 10.4 min) as the mixture of diastereomers (**Fig. 7(a)**). As one could expect from the solution phase experiments, **8** surprisingly did not undergo extensive loss of the phosphoramidate linkage on solid phase by I₂ in Py/H₂O method.

The trimer **8** suffered ca. 7% loss of the phosphoramidate linkages, while the major fraction was eluted at retention time 10.2 and 10.4 min (**Fig. 7(b)**).

Presently, a number of 2'-modifications that offer the dual advantage of high nuclease resistance and enhanced binding affinity have been reported.¹⁴ The 2'-O-(2-methoxyethyl) (2'-O-MOE),¹⁵ with the phosphodiester (PO) backbone provides sufficient nuclease resistance, at approximately the same level as the 2'-O-deoxyphosphorothioate modification. A typical approach entails the use of the 2'-O-alkylated phosphodiester to replace part of the PS linkers in the antisense compounds, leading to a MBO with reduced phosphorothioate content. The preliminary experiments have shown that these MBOs exhibit more acceptable safety profiles than those of the PS oligos.¹⁶ As a consequence of which, the antisense oligos containing PS and PO linkages in the backbone are presently undergoing human clinical trials.¹⁷ Taking into account the superior attributes of the developed CCl₄/H₂O/NMM/Py/CH₃CN (2.5/0.2/1.0/6.0/9.3, v/v) oxidation method and the importance of the MBOs containing both phosphorothioate and phosphodiester linkages, we have extended our developed methodology to the synthesis of d(Tp_oTp_sTp_sTp_sTp_sTp_sT) (**9**). The expected product was obtained at retention time 31.0 min (**Fig. 8**), which was simultaneously characterized by ESI-MS (shown in the inset of **Fig. 8**). The LC-MS profile of **9** indicated that the overall desulfurization was less than 4% (desulfurized **8** mer oligo retention time 30.35 min, containing 2 phosphodiester and 5 phosphorothioate linkages in the backbone).

Conclusion

In the present study we have developed $\text{CCl}_4/\text{H}_2\text{O}/\text{NMM}/\text{Py}/\text{CH}_3\text{CN}$ mixture as a mild and efficient oxidizing reagent, that can quantitatively convert nucleoside H-phosphonate diesters to the corresponding phosphates both in the solution phase as well as in the solid phase synthesis of oligonucleotides. The data presented suggests that this oxidation method can serve as an advantageous replacement for the conventionally used aqueous iodine in pyridine oxidation method, particularly in the synthesis of Mixed Backbone Oligonucleotides.

Experimental

Materials

All the nucleoside H-phosphonates, Icaac CPG-500 Å were purchased from Glen Res. (Virginia). Acetonitrile (HPLC grade, Spectrochem India Ltd.) was distilled and stored over activated 3 Å molecular sieves (Aldrich Chemical Co.). Pyridine (E. Merck, India) was dried by refluxing over KOH and was stored over activated 3 Å molecular sieves. Pivaloyl chloride (AR grade, Spectrochem India Ltd.) was used as received. The solid phase oligonucleotide syntheses were performed manually in a 2.5 mL Hamilton gas tight syringe with a glass wool plug at the inlet.¹⁸

Methods

³¹P NMR analysis: the ³¹P NMR was performed on a Bruker AMX 500 instrument. The experiments were carried out in an NMR sample tube (5 mm×180 mm) at 25 °C with proton decoupling at 202 Mz for ³¹P NMR using 85% H_3PO_4 as external reference. In each measurement, data accumulation was carried out 64 times with a pulse width of 10 μs, an acquisition time of 0.196 s, and a pulse delay of 3.0 s. The spectrum was acquired with 16384 data points for the spectral width of 41666 Hz.

RP-HPLC techniques: the RP-HPLC analysis of oligonucleotides **3** along with the positive and negative control experiments were performed on a Licrosphere RP-18 (e) 5 μm, (125×4 mm) column using a Merck Hitachi LC pump L-7100 which was connected to a diode array detector (model L-7455). The system used was A: 0.1 M TEAA, pH=7.0, B: 30% CH_3CN in 0.1 M TEAA (pH=7.0) using a gradient A-B in 20 min, flow rate: 1.5 mL min⁻¹. RP-HPLC analysis of oligonucleotides **7** and **8** were performed on a HP spherisorb ODS-2, 5 μm, (125×4 mm) column using a Varian Prostar LC pump to which was attached a Dynamax absorbance detector (model UV-D11). The solvent system and the conditions were same as described above. The RP-HPLC analysis of the oligonucleotides **4** and **9** were performed on a Waters YMC-PackTM ODS-AQTM S-5 column (2.0×250 mm) using a Waters HPLC system (600 E system controller, 996 Photo-

diode Array Detector). All these RP-HPLC fractions were simultaneously characterized by electrospray ionization mass spectrometry (ESI-MS) using a Hewlett-Packard 59987A electrospray quadrupole mass analyzer using negative polarity.

Solution phase synthesis of dithymidyl H-phosphonate diester (**1**)

To a well stirred solution of dT-H-phosphonate TEA salt (0.1 mmol) and 3'-O-acetyl thymidine (0.11 mmol) dissolved in 1.0 mL of anhydrous pyridine, was added pivaloyl chloride (0.1 mmol). The reaction mixture was stirred at ambient temperature for 15 min. The reaction mixture was monitored by TLC (ethyl acetate/pet ether=8:2). After completion of the reaction, the reaction mixture was concentrated and was subjected to column chromatography to get the product **1** (isolated yield 85%).

³¹P NMR (CD_3CN) **1**, δ = 12.1 ppm.

Solution phase synthesis of dithymidyl phosphonate (**2**) using $\text{CCl}_4/\text{H}_2\text{O}/\text{NMM}/\text{Py}/\text{CH}_3\text{CN}$ oxidation method

The dithymidyl H-phosphonate diester **1** (10 mg, ca. 0.01 mmol) δ = 12.1 ppm) was dissolved in CD_3CN (200 μL) and was transferred into a NMR sample tube. The 2.0 mL of $\text{CCl}_4/\text{H}_2\text{O}/\text{NMM}/\text{Py}/\text{CH}_3\text{CN}$ (2.5/1.0/1.0/6.0/1.0, v/v) mixture was added at ambient temperature to the above solution. The reaction was monitored by ³¹P NMR. After 5 min, quantitative formation of **2** (δ = 2.9 ppm) was observed as the only detectable product.

Solution phase synthesis of **2** using 0.035 M I_2 in Py/THF/ H_2O (1/300/33, v/v) oxidation method

The solution of **1** (10 mg, ca. 0.01 mmol) in CD_3CN (200 μL) was transferred into a NMR sample tube. To this solution was added 2.0 mL of 0.035 M I_2 in Py/THF/ H_2O (1/300/33, v/v) at ambient temperature, the progress of the reaction was monitored by ³¹P NMR spectroscopy. After 5 min formation of **2** (δ = 3.0 ppm) along with the formation of another product at δ = -9.8 ppm was detected.

Solution phase synthesis of **2** using 0.2 M I_2 in Py/ H_2O (98:2, v/v) oxidation method

As previously described, to the solution of **1** (10 mg, ca. 0.01 mmol) dissolved in CD_3CN (200 μL) was added 2.0 mL of 0.2 M I_2 in Py/ H_2O (98:2, v/v). After 5 min quantitative formation of **2** (δ = 2.5 ppm) was observed as the only detectable product.

Synthesis of oligodeoxynucleotides (ODNs)

Solid phase synthesis of d(AA) (**3**) using $\text{CCl}_4/\text{H}_2\text{O}/\text{NMM}/\text{Py}/\text{CH}_3\text{CN}$ oxidation method (protocol 1).

The DMT-A^{bz} loaded Icaac CPG 500 Å (22 mg, 1 μmole, loading 44.3 μmol/g) was poured into a well dried 2.5 mL Hamilton gas tight syringe with a glass wool

plug at the inlet. The support was initially washed with CH_3CN (3×0.5 mL). The DMT protecting group was removed by washing with dichloroacetic acid (DCA)/ CH_2Cl_2 (3/100, v/v, 1.0 mL). Additional washes with DCA/ CH_2Cl_2 (8×0.5 mL) were performed and all the orange effluents were collected for subsequent spectroscopy (448 nm) and calculation of the coupling efficiency. The support was washed successively with Py/ CH_3CN (1/4, v/v, 1×0.5 mL) and dry CH_3CN (4×0.5 mL). The dA^{bz} -H-phosphonate solution (10 mg, ca. 0.012 mmol) in dry CH_3CN (0.25 mL) and dry pyridine (0.25 mL) containing pivaloyl chloride (8 μL , ca. 0.075 mmole) was drawn into the syringe and was agitated for 2 min. The coupling agents were ejected from the syringe, and the support was washed with dry CH_3CN (3×0.5 mL). The internucleoside H-phosphonate linkages were then oxidized with 2.0 mL of $\text{CCl}_4/\text{H}_2\text{O}/\text{NMM}/\text{Py}/\text{CH}_3\text{CN}$ (2.5/0.2/1.0/6.0/9.3, v/v) for 45 min. Washings were performed with Py/ CH_3CN (1/4, v/v, 3×1.5 mL) and dry CH_3CN (3×1.5 mL). The terminal DMT protecting group was removed by washing with dichloroacetic acid (DCA)/ CH_2Cl_2 (3/100, v/v, 1.0 mL) and additional washes with DCA/ CH_2Cl_2 (8×0.5 mL) followed by subsequent washings with Py/ CH_3CN (1/4, v/v, 1×0.5 mL) and dry CH_3CN (4×0.5 mL). The support bound oligonucleotide was then completely deprotected using concentrated NH_4OH , 55°C , 16 h.

RP-HPLC of **3** (retention time 7.76 min).

^{31}P NMR (D_2O) **3**, $\delta = 0.9$ ppm.

The negative control experiment was performed exactly identical to protocol 1. The only difference was that the oxidation step was omitted. The support bound oligonucleotide was then completely deprotected using concentrated NH_4OH , 55°C , 16 h.

RP-HPLC of **3** (retention time 7.79 min), adenosine (retention time 5.95 min), dA -H-phosphonate (retention time 6.4 min).

^{31}P NMR (D_2O) dA -H-phosphonate, $\delta = 6.9$ and 8.3 ppm.

Solid phase synthesis of **3** using 0.2 M I_2 in Py/ H_2O (98:2, v/v) oxidation method (positive control)

The dimer was synthesized by coupling dA^{bz} -H-phosphonate with the support bound dA^{bz} loaded 1caa CPG 500 Å as described in protocol 1. At the end of the synthesis, before the final detritylation, the internucleoside H-phosphonate linkages were oxidized with 2.0 mL of 0.2 M solution in Py/ H_2O (98:2, v/v) for 30 min. The support was then washed with $\text{CH}_3\text{CN}/\text{Py}$ (4/1, v/v, 6×1.5 mL) and with CH_3CN (3×1.5 mL). After the removal of the DMT group, the support bound oligonucleotide was then completely deprotected using concentrated NH_4OH , 55°C , 16 h.

RP-HPLC of **3** (retention time 7.76 min), dA -H phosphonate (retention time 6.4 min).

^{31}P NMR (D_2O) **3**, $\delta = 0.9$ ppm.

Solid phase synthesis of **d(GTTAAGACTTTTAC)** (**4**) using $\text{CCl}_4/\text{H}_2\text{O}/\text{NMM}/\text{Py}/\text{CH}_3\text{CN}$ oxidation method

The DMT- C^{bz} loaded 1caa CPG 500 Å (21 mg, 1 μmole , loading 46.9 ($\mu\text{mol/g}$) was poured into a well dried 2.5 mL Hamilton gas tight syringe with a glass wool plug at the inlet. The support was initially washed with CH_3CN (3×0.5 mL). The DMT protecting group was removed by washing with dichloroacetic acid (DCA)/ CH_2Cl_2 (3/100, v/v, 1.0 mL). Additional washes with DCA/ CH_2Cl_2 (8×0.5 mL) were performed and all the orange effluents were collected for subsequent spectroscopy (448 nm) and calculation of the coupling efficiency. The support was then washed with Py/ CH_3CN (1/4, v/v, 1×0.5 mL) and dry CH_3CN (4×0.5 mL). The dA^{bz} -H-phosphonate solution (10 mg, ca. 0.012 mmol) in dry CH_3CN (0.25 mL) and dry pyridine (0.25 mL) containing pivaloyl chloride (8 L, ca. 0.075 mmole) was drawn into the syringe and was agitated for 2 min. The coupling agents were ejected from the syringe, and the support was washed with dry CH_3CN (3×0.5 mL). The terminal DMT protecting group was removed by washing with DCA/ CH_2Cl_2 (3/100, v/v, 1.0 mL) and additional washes with DCA/ CH_2Cl_2 (8×0.5 mL) followed by subsequent washings with Py/ CH_3CN (1/4, v/v, 1×0.5 mL) and dry CH_3CN (4×0.5 mL). This cycle was repeated by coupling of the appropriate nucleosidic unit, until the desired length of the oligonucleotide was achieved. The oxidation step was then carried out with 2.0 mL of $\text{CCl}_4/\text{H}_2\text{O}/\text{NMM}/\text{Py}/\text{CH}_3\text{CN}$ (2.5/0.2/1.0/6.0/9.3, v/v) for 45 min at the end of the synthesis and before final detritylation. After the final detritylation and washings steps, the oligonucleotide was completely deprotected using concentrated NH_4OH , 55°C , 24 h.

RP-HPLC of **4** (retention time 27.8 min).

ESI-MS of **4**, $M_{\text{w found}}$: 4234.8 $M_{\text{w calcd}}$: 4237.

Solution phase stability of **d(TpsT)** (**5**) in $\text{CCl}_4/\text{H}_2\text{O}/\text{NMM}/\text{Py}/\text{CH}_3\text{CN}$ oxidizing system

The DMT-T loaded 1caa CPG 500 Å (20 mg, 1 μmol , loading 48.3 $\mu\text{mol/g}$) was poured into a well dried 2.5 mL Hamilton gas tight syringe with a glass wool plug at the inlet. The support was initially washed with CH_3CN (3×0.5 mL). The DMT protecting group was removed by washing with DCA/ CH_2Cl_2 (3/100, v/v, 1.0 mL). Additional washes with DCA/ CH_2Cl_2 (8×0.5 mL) were performed and all the orange effluents were collected for subsequent spectroscopy (448 nm) and calculation of the coupling efficiency. The support was washed successively with Py/ CH_3CN (1/4, v/v, 1×0.5 mL) and dry CH_3CN (4×0.5 mL). The dT -H-phosphonate solution (10 mg, ca. 0.015 mmol) in dry CH_3CN (0.25 mL) and dry pyridine (0.25 mL) containing pivaloyl chloride (8 μL , ca. 0.075 mmol) was drawn into the syringe and was agitated for 2 min. The coupling agents were ejected from the syringe, and the support was washed with dry CH_3CN (3×0.5 mL). The internucleotidic H-phosphonate linkages were sulfurized by using 2.0 mL mixture of 10% S_8 in $\text{CS}_2/\text{Py}/\text{TEA}$ (35/35/1) for

3 h. Washings were performed with Py/CH₃CN (1/4, v/v, 6×1.5 mL) and dry CH₃CN (3×1.5 mL). The terminal DMT protecting group was removed by washing with DCA/CH₂Cl₂ (3/100, v/v, 1.0 mL) and additional washes with DCA/CH₂Cl₂ (8×0.5 mL) followed by subsequent washings with Py/CH₃CN (1/4, v/v, 1×0.5 mL) and dry CH₃CN (4×0.5 mL). After performing the deprotection with concentrated NH₄OH, 5°C, 16 h, the oligo was purified by RP-HPLC. The purified oligo 5 (1.0 A₂₆₀ units) was then stirred in a Wheaton glass vial for 16 h at ambient temperature with 1.0 mL of CCl₄/H₂O/NMM/Py/CH₃CN (2.5/1.0/1.0/6.0/1.0, v/v) mixture. The reaction mixture was lipholyzed and was then analyzed by ³¹P NMR spectroscopy.

³¹P NMR (D₂O) 5, δ = 57.4 and 57.0 ppm; desulfurized dimer d(TpoT) δ = 0.9 ppm.

Solution phase stability of 5 in 0.2 M I₂ in Py/H₂O (98:2, v/v) oxidizing system

As described above the purified oligo 5 (1.0 A₂₆₀ units) was stirred in a Wheaton glass vial for 16 h at ambient temperature with 1.0 mL of 0.2 M I₂ in Py/H₂O (98:2, v/v) mixture. After lipholyzation the ³¹P NMR analysis was performed.

³¹P NMR (D₂O) desulfurized dimer d(TpoT) = 0.9 ppm

Solution phase stability of d(T_{PN}T) (6) in CCl₄/H₂O/NMM/Py/CH₃CN oxidizing system

The DMT-T loaded 1caa CPG 500 Å (20 mg, 1 μmol, loading 48.3 (μmol/g) was poured into a well dried 2.5 mL Hamilton gas tight syringe with a glass wool plug at the inlet. The support was initially washed with CH₃CN (3×0.5 mL). The DMT protecting group was removed by washing with DCA/CH₂Cl₂ (3/100, v/v, 1.0 mL). Additional washes with DCA/CH₂Cl₂ (8×0.5 mL) were performed and all the orange effluents were collected for subsequent spectroscopy (448 nm) and calculation of the coupling efficiency. The support was washed successively with Py/CH₃CN (1/4, v/v, 1×0.5 mL) and dry CH₃CN (4×0.5 mL). The dT-H-phosphonate solution (10 mg, ca. 0.015 mmol) in dry CH₃CN (0.25 mL) and dry pyridine (0.25 mL) containing pivaloyl chloride (8 μL, ca. 0.075 mmol) was drawn into the syringe and was agitated for 2 min. The coupling agents were ejected from the syringe, and the support was washed with dry CH₃CN (3×0.5 mL). The dithymidyl H-phosphonate diester linkages were converted to the phosphoramidate linkages by using 2.0 mL mixture of 15% *N,N*-dimethyl-1,3-propane diamine in CCl₄ for 1 h. Washings were performed with Py/CH₃CN (1/4, v/v, 3×1.5 mL) and dry CH₃CN (3×1.5 mL). The terminal DMT protecting group was removed by washing with DCA/CH₂Cl₂ (3/100, v/v, 1.0 mL) and additional washes with DCA/CH₂Cl₂ (8×0.5 mL) followed by washings with Py/CH₃CN (1/4, v/v, 1×0.5 mL) and dry CH₃CN (4×0.5 mL). After performing the deprotection with concentrated NH₄OH, 55°C, 16 h, the oligo was

purified by RP-HPLC. The purified oligo 5 (1.0 A₂₆₀ units) was then stirred in a Wheaton glass vial for 16 h, at ambient temperature with 1.0 mL of CCl₄/H₂O/NMM/Py/CH₃CN (2.5/1.0/1.0/6.0/1.0, v/v) mixture. The ³¹P NMR analysis was carried out after lipholyzation of the reaction mixture.

³¹P NMR (D₂O) 6, δ = 12.08 ppm and 12.02 ppm.

Solution phase stability of 6 in 0.2 M I₂ in Py/H₂O (98:2, v/v) oxidizing system

As described above, the purified oligo 6 (1.0 A₂₆₀ units) was stirred with 1.0 mL of 0.2 M I₂ in Py/H₂O (98:2, v/v) mixture in a Wheaton glass vial for 16 h at ambient temperature. The lipholyzed reaction mixture was subjected to ³¹P NMR analysis.

³¹P NMR (D₂O) desulfurized dimer d(TpoT) δ = 0.97 ppm.

Solid phase synthesis of d(TpoTpsT) (7) using CCl₄/H₂O/NMM/Py/CH₃CN oxidation method

The DMT-T loaded 1caa CPG 500 Å (20 mg, 1 μmol, loading 48.3 (μmol/g) was poured into a well dried 2.5 mL Hamilton gas tight syringe with a glass wool plug at the inlet. The support was initially washed with CH₃CN (3×0.5 mL). The DMT protecting group was removed by washing with DCA/CH₂Cl₂ (3/100, v/v, 1.0 mL). Additional washes with DCA/CH₂Cl₂ (8×0.5 mL) were performed and all the orange effluents were collected for subsequent spectroscopy (448 nm) and calculation of the coupling efficiency. The support was washed successively with Py/CH₃CN (1/4, v/v, 1×0.5 mL) and dry CH₃CN (4×0.5 mL). The dT-H-phosphonate solution (10 mg, ca. 0.015 mmol) in dry CH₃CN (0.25 mL) and dry pyridine (0.25 mL) containing pivaloyl chloride (8 μL, ca. 0.075 mmol) was drawn into the syringe and was agitated for 2 min. The coupling agents were ejected from the syringe, and the support was washed with dry CH₃CN (3×0.5 mL). The internucleotidic H-phosphonate linkages were sulfurized by using 2.0 mL mixture of 10% S₈ in CS₂/Py/TEA (35/35/1) for 3 h. Washings were performed with Py/CH₃CN (1/4, v/v, 6×1.5 mL) and dry CH₃CN (3×1.5 mL). The terminal DMT protecting group was removed by washing with DCA/CH₂Cl₂ (3/100, v/v, 1.0 mL) and additional washes with DCA/CH₂Cl₂ (8×0.5 mL), followed by subsequent washings with Py/CH₃CN (1/4, v/v, 1×0.5 mL) and dry CH₃CN (4×0.5 mL). The next coupling step was then effected by drawing the dT-H-phosphonate solution (10 mg, ca. 0.015 mmol) in dry CH₃CN (0.25 mL) and dry pyridine (0.25 mL) containing pivaloyl chloride (8 μL, ca. 0.075 mmol) into the syringe and agitating the mixture for 2 min. The coupling agents were ejected from the syringe, and the support was washed with dry CH₃CN (3×0.5 mL). The internucleoside H-phosphonate linkages were then oxidized with 2.0 mL of CCl₄/H₂O/NMM/Py/CH₃CN (2.5/0.2/1.0/6.0/9.3, v/v) for 45 min. Washings were performed with Py/CH₃CN (1/4, v/v, 3×1.5 mL) and dry CH₃CN (3×1.5 mL). The terminal DMT protecting

group was removed by washing with DCA/CH₂Cl₂ (3/100, v/v, 1.0 mL) and additional washes with DCA/CH₂Cl₂ (8×0.5 mL) followed by subsequent washings with Py/CH₃CN (1/4, v/v, 1×0.5 mL) and dry CH₃CN (4×0.5 mL). The support bound oligonucleotide was then completely deprotected using concentrated NH₄OH, 55 °C, 16 h.

RP-HPLC of **7** (retention time 9.8 and 10.3 min); desulfurized trimer d(TpoTpoT) retention time 9.1 min.

Solid phase synthesis of **7** using 0.2 M I₂ in Py/H₂O (98:2, v/v) oxidation method

The synthesis of **7** was carried out identically as for the synthesis described above by the CCl₄/H₂O/NMM/Py/CH₃CN oxidation method. The only difference was that the oxidation reaction was carried out at the end of the synthesis, before final detritylation by 2.0 mL of 0.2 M I₂ in Py/H₂O (98:2, v/v) method for 30 min instead of CCl₄/H₂O/NMM/Py/CH₃CN method. The support was then washed with CH₃CN/Py (4/1, v/v, 6×1.5 mL) and with CH₃CN (3×1.5 mL). After the removal of the DMT group, the support bound oligonucleotide was then completely deprotected using concentrated NH₄OH, 55 °C, 16 h.

RP-HPLC of **7** (retention time 9.9 and 10.3 min), desulfurized trimer d(TpoTpoT) (retention time 9.1 min).

Solid phase synthesis of d(TpoTPNT) (**8**) using CCl₄/H₂O/NMM/Py/CH₃CN oxidation method

The DMT-T loaded 1caa CPG 500 Å (20 mg, 1 µmol, loading 48.3 µmol/g) was poured into a well dried 2.5 mL Hamilton gas tight syringe with a glass wool plug at the inlet. The support was initially washed with CH₃CN (3×0.5 mL). The DMT protecting group was removed by washing with DCA/CH₂Cl₂ (3/100, v/v, 1.0 mL). Additional washes with DCA/CH₂Cl₂ (8×0.5 mL) were performed and all the orange effluents were collected for subsequent spectroscopy (448 nm) and calculation of the coupling efficiency. The support was washed successively with Py/CH₃CN (1/4, v/v, 1×0.5 mL) and dry CH₃CN (4×0.5 mL). The dT-H-phosphonate solution (10 mg, ca. 0.015 mmol) in dry CH₃CN (0.25 mL) and dry pyridine (0.25 mL) containing pivaloyl chloride (8 µL, ca. 0.075 mmol) was drawn into the syringe and was agitated for 2 min. The coupling agents were ejected from the syringe, and the support was washed with dry CH₃CN (3×0.5 mL). The dithymidyl H-phosphonate diester linkages were converted to the phosphoramidate linkages by using 2.0 mL mixture of 15% *N,N*-dimethyl-1,3-propane diamine in CCl₄ for 1 h. Washings were performed with Py/CH₃CN (1/4, v/v, 3×1.5 mL) and dry CH₃CN (3×1.5 mL). The terminal DMT protecting group was removed by washing with DCA/CH₂Cl₂ (3/100, v/v, 1.0 mL) and additional washes with DCA/CH₂Cl₂ (8×0.5 mL) followed by subsequent washings with Py/CH₃CN (1/4, v/v, 1×0.5 mL) and dry CH₃CN (4×0.5 mL). The next coupling step was then effected by drawing the dT-H-phosphonate solution (10 mg, ca.

0.015 mmol) in dry CH₃CN (0.25 mL) and dry pyridine (0.25 mL) containing pivaloyl chloride (8 µL, ca. 0.075 mmol) into the syringe and agitating the mixture for 2 min. The coupling agents were ejected from the syringe, and the support was washed with dry CH₃CN (3×0.5 mL). The internucleoside H-phosphonate linkages were then oxidized with 2.0 mL of CCl₄/H₂O/NMM/Py/CH₃CN (2.5/0.2/1.0/6.0/9.3, v/v) for 45 min. Washings were performed with Py/CH₃CN (1/4, v/v, 3×1.5 mL) and dry CH₃CN (3×1.5 mL). The terminal DMT protecting group was removed by washing with DCA/CH₂Cl₂ (3/100, v/v, 1.0 mL) and additional washes with DCA/CH₂Cl₂ (8×0.5 mL) followed by subsequent washings with Py/CH₃CN (1/4, v/v, 1×0.5 mL) and dry CH₃CN (4×0.5 mL). The support bound oligonucleotide was then completely deprotected using concentrated NH₄OH, 55 °C, 16 h.

RP-HPLC of **8** (retention time 10.2 and 10.4 min); desulfurized trimer d(TpoTpoT) (retention time 9.08 min).

Solid phase synthesis of **8** using 0.2 M I₂ in Py/H₂O (98:2, v/v) oxidation method

The synthesis of **8** was carried out identically as for the synthesis described above by the CCl₄/H₂O/NMM/Py/CH₃CN oxidation method. The only difference was that the oxidation reaction was carried out at the end of the synthesis, before final detritylation by 2.0 mL of 0.2 M I₂ in Py/H₂O (98:2, v/v) method for 30 min instead of CCl₄/H₂O/NMM/Py/CH₃CN method. The support was then washed with CH₃CN/Py (4/1, v/v, 6×1.5 mL) and with CH₃CN (3×1.5 mL). After the removal of the DMT group, the support bound oligonucleotide was then completely deprotected using concentrated NH₄OH, 55 °C, 16 h.

RP-HPLC of **8** (retention time 10.2 and 10.4 min), desulfurized trimer d(TpoTpoT) (retention time 9.04 min).

Synthesis of d(TpoTpsTpsTpsTpsTpsTpsT) (**9**) using CCl₄/H₂O/NMM/Py/CH₃CN oxidation method

The DMT-T loaded 1caa CPG 500 Å (20 mg, 1 µmol, loading 48.3 µmol/g) was poured into a well dried 2.5 mL Hamilton gas tight syringe with a glass wool plug at the inlet. The support was initially washed with CH₃CN (3×0.5 mL). The DMT protecting group was removed by washing with DCA/CH₂Cl₂ (3/100, v/v, 1.0 mL). Additional washes with DCA/CH₂Cl₂ (8×0.5 mL) were performed and all the orange effluents were collected for subsequent spectroscopy (448 nm) and calculation of the coupling efficiency. The support was washed successively with Py/CH₃CN (1/4, v/v, 1×0.5 mL) and dry CH₃CN (4×0.5 mL). The dT-H-phosphonate solution (10 mg, ca. 0.015 mmol) in dry CH₃CN (0.25 mL) and dry pyridine (0.25 mL) containing pivaloyl chloride (8 µL, ca. 0.075 mmol) was drawn into the syringe and was agitated for 2 min. The coupling agents were ejected from the syringe, and the support was washed with dry CH₃CN (3×0.5 mL). This cycle was repeated

until the oligonucleotide, d(T_{PH}T_{PH}T_{PH}T_{PH}T_{PH}T_{PH}T) (where PH corresponds to the H-phosphonate diester linkages in the backbone) was achieved. The internucleotidic H-phosphonate linkages were sulfurized by using 2.0 mL mixture of 10% S₈ in CS₂/PyTEA (35/35/1) for 3 h. Washings were performed with Py/CH₃CN (1/4, v/v, 6×1.5 mL) and dry CH₃CN (3×1.5 mL). The terminal DMT protecting group was removed by washing with dichloroacetic acid (DCA)/CH₂Cl₂ (3/100, v/v, 1.0 mL) and additional washes with DCA/CH₂Cl₂ (8×0.5 mL) followed by subsequent washings with Py/CH₃CN (1/4, v/v, 1×0.5 mL) and dry CH₃CN (4×0.5 mL). The last dT nucleosidic unit was added by drawing the dT-H-phosphonate solution (10 mg, ca. 0.015 mmol) in dry CH₃CN (0.25 mL) and dry pyridine (0.25 mL) containing pivaloyl chloride (8 µL, ca. 0.075 mmol) into the syringe and agitating the mixture for 2 min. The coupling agents were ejected from the syringe, and the support was washed with dry CH₃CN (3×0.5 mL). The internucleoside H-phosphonate linkage was then oxidized with 2.0 mL of CCl₄/H₂O/NMM/Py/CH₃CN (2.5/0.2/1.0/6.0/9.3, v/v) for 45 min. Washings were performed with Py/CH₃CN (1/4, v/v, 3×1.5 mL) and dry CH₃CN (3×1.5 mL). The terminal DMT protecting group was removed by washing with DCA/CH₂Cl₂ (3/100, v/v, 1.0 mL) and additional washes with DCA/CH₂Cl₂ (8×0.5 mL) followed by subsequent washings with Py/CH₃CN (1/4, v/v, 1×0.5 mL) and dry CH₃CN (4×0.5 mL). The support bound oligonucleotide was then completely deprotected using concentrated NH₄OH, 55 °C, 16 h.

RP-HPLC of **9** (retention time 31.0 min).

ESI-MS spectrum of **9**, M_w_{calc.}: 2459.95 M_w_{found.}: 2466.

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